

SHORT  
COMMUNICATIONS

## Aryl 2,3-Dibromopropyl Sulfones in S,N-Tandem Heterocyclizations. New Synthesis of Benzimidazothiazolidines

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In the recent years, extensive development in the chemistry of saturated heterocycles containing nitrogen and sulfur atoms, specifically thiazine and thiazolidine derivatives, resulted primarily from their high pharmacological activity [1–3]. Thiazolidine ring constitutes a structural fragment of many biologically important compounds, in particular penicillin derivatives. However, methods of synthesis of such heterocycles are few in number, multistep, and fairly laborious [4–7]. We now propose a convenient procedure for synthesizing benzimidazothiazolidines via S,N-tandem alkylation of 2-sulfanylbenzimidazole with aryl 2,3-dibromopropyl sulfones (Scheme 1).

The reactions were carried out by stirring a mixture of the corresponding sulfone **Ia–Id**, 2-sulfanylbenzimidazole, and potassium hydroxide (reactant molar ratio 1:2:4) in ethanol at 20–25°C (reaction time 8 h). The yields of products **IIa–IId** were 85–95%. Their structure was proved by <sup>1</sup>H NMR spectroscopy and mass spectrometry. Compounds **IIa–IId** showed in the <sup>1</sup>H NMR spectra signals from protons of the CH<sub>2</sub>CHCH<sub>2</sub> group as an AA'MM'X spin system. Nonequivalence of the CH<sub>2</sub>S protons confirms the formation of thiazolidine ring having two chiral centers. The mass spectra of **IIa–IId** contained the following characteristic fragment ion peaks, *m/z*: 188

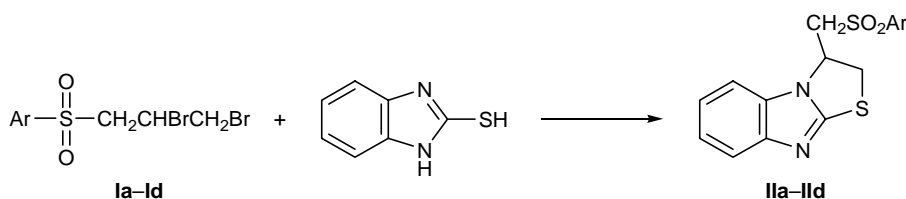
[*M* – ArS(O)OH]<sup>+</sup>, 175 [*M* – ArSO<sub>2</sub>CH<sub>2</sub>]<sup>+</sup>, 150, [*M* – ArSO<sub>2</sub>CH=CH<sub>2</sub>]<sup>+</sup>, 130 [*M* – ArSO<sub>2</sub>CH<sub>2</sub>–CH=S]<sup>+</sup>. The fragmentation pattern was proposed with account taken of the data in [8, 9]. The absence of isomeric compounds among the products indicates high chemo- and regioselectivity of the reaction, typical of tandem processes [10].

**3-Phenylsulfonylmethyl-2,3-dihydro[1,3]thiazolo[3,2-*a*]benzimidazole (IIa)**. Yield 90%, mp 162–163°C (from ethanol). <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>), δ, ppm: 3.27 d (1H, CH), 3.56 d.d (1H, CH), 4.09 d (1H, CH), 4.26 d.d (1H, CH), 5.24 t (1H, CH), 7.01–7.96 m (9H, H<sub>arom</sub>). Mass spectrum, *m/z*, (*I*<sub>rel</sub>, %): 330 (66.4) [*M*]<sup>+</sup>, 188 (100), 175 (50.2), 150 (8.2), 130 (16.4), 117 (24.5), 90 (17.2).

**3-(4-Methylphenylsulfonylmethyl)-2,3-dihydro[1,3]thiazolo[3,2-*a*]benzimidazole (IIb)**. Yield 95%, mp 174–175°C (from ethanol). <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>), δ, ppm: 2.46 s (3H, CH<sub>3</sub>), 3.26 d (1H, CH), 3.54 d.d (1H, CH), 4.07 d (1H, CH), 4.26 d.d (1H, CH), 5.21 t (1H, CH), 7.39 d (2H, H<sub>arom</sub>), 7.55 d (2H, H<sub>arom</sub>). Mass spectrum, *m/z*, (*I*<sub>rel</sub>, %): 344 (76.6) [*M*]<sup>+</sup>, 188 (100), 175 (54.6), 150 (9.8), 130 (17.1), 117 (25.2), 90 (16.8).

**3-(4-Nitrophenylsulfonylmethyl)-2,3-dihydro[1,3]thiazolo[3,2-*a*]benzimidazole (IIc)**. Yield 92%,

Scheme 1.



Ar = Ph (**a**), 4-MeC<sub>6</sub>H<sub>4</sub> (**b**), 4-O<sub>2</sub>NC<sub>6</sub>H<sub>4</sub> (**c**), 2-naphthyl (**d**).

mp 238°C (from ethanol). <sup>1</sup>H NMR spectrum (DMSO-*d*<sub>6</sub>), δ, ppm: 3.91 d (1H, CH), 4.09 d (1H, CH), 4.12–4.15 m (1H, CH), 4.30 t (1H, CH), 5.29 m (1H, CH), 7.04–7.29 m (4H, H<sub>arom</sub>), 8.17 d (2H, H<sub>arom</sub>), 8.29 d (2H, H<sub>arom</sub>). Mass spectrum, *m/z*, (*I*<sub>rel</sub>, %): 375 (78.3) [*M*]<sup>+</sup>, 188 (100), 175 (64.5), 150 (8.8), 130 (18.8), 117 (22.7), 90 (15.6).

**3-(2-Naphthylsulfonylmethyl)-2,3-dihydro[1,3]-thiazolo[3,2-*a*]benzimidazole (II<sub>d</sub>).** Yield 85%, mp 167–168°C (from ethanol). <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>), δ, ppm: 3.38 d (1H, CH), 3.63 d.d (1H, CH), 4.14–4.18 m (2H, 2CH), 5.29 t (1H, CH), 7.04–7.56 m (4H, H<sub>arom</sub>), 7.67–8.03 m (7H, H<sub>arom</sub>). Mass spectrum, *m/z*, (*I*<sub>rel</sub>, %): 380 (100) [*M*]<sup>+</sup>, 188 (80.1), 175 (79.2), 150 (8.4), 130 (21.1), 117 (36.4), 90 (18.2).

The GC–MS data were obtained on a Micromass ZDM-2000 instrument (electrospray ionization, positive ions). The mass spectra (70 eV) were run on an MKh-1321 spectrometer with direct sample admission into the ion source. The <sup>1</sup>H NMR spectra were measured from solutions in DMSO-*d*<sub>6</sub> and CDCl<sub>3</sub> on a Bruker AM-500 spectrometer (500.13 MHz) using signals from residual protons in the solvents as internal reference.

## REFERENCES

1. Yoshii, M., *Farumashia*, 2000, vol. 2, p. 442.
2. Kobayashi, T., *Molecular Medicine*, 2001, vol. 2, p. 196.
3. Roth, H.J. and Fenner, H., *Struktur-Bioreaktivität-Wirkungsbezogene Eigenschaften Deutscher Apotheker*, Stuttgart, 2000, p. 441.
4. Thyagarajan, B.S. and Glowienka, J.A., *Phosphorus Sulfur*, 1988, vol. 39, p. 11.
5. Uanefeld, W. and Schlitzer, M., *J. Heterocycl. Chem.*, 1995, vol. 32, p. 1019.
6. Weyler, S., Hayallah, M.A., and Muller, C.E., *Tetrahedron*, 2003, vol. 59, p. 47.
7. Choudhury, A.Z., Shaifullah, M., Shibata, Y., Marita, M., Kaya, K., and Hiratani, K., *J. Heterocycl. Chem.*, 2001, vol. 38, p. 1173.
8. Ponomarev, D.A. and Takhistov, V.V., *Recent Advances in Analytical Techniques*, Attaur-Rahman, M., Ed., Reading: Hardwood Academic, 2002, p. 369.
9. Ponomarev, D.A., Golovin, A.V., and Takhistov, V.V., *Eur. J. Mass Spectrom.*, 2002, vol. 8, p. 409.
10. Nicolau, K.C., Tamsyn Montagu, and Scott A. Snyder, *Chem. Commun.*, 2003, p. 551.